LT100 GL Sciences Inc.

HPLC has become the dominant analytical separation tools in all industries and laboratories. Numerous HPLC columns are commercially available in the market, but it is difficult to choose the correct HPLC column for your application needs. In the previous note (No. 50), we introduced various chromatograms comparing several reversed phase Inertsil HPLC columns, which was well-received.

Since then we released several new HPLC columns, such as InertSustain C18. Hence, this note is an updated version to the previous note which all reversed phase Inertsil columns are shown and compared again.

(C. Aoyama)

Columns compared in this note

Column name	Functional group	Features			
InertSustain C18	$\neg \land \land$	InertSustain C18 inherits the advantages of all the current Inertsil HPLC columns, but now can be used for wide pH analysis with consistent performance.			
Inertsil ODS-4		It is excellent at inertness. In general, elution time of Inertsil ODS-4 is shorter than InertSustain C18 and ODS-3.			
Inertsil ODS-3		An ODS column released in 1994, but still is heavily used all over the world because of its strong hydrophobic property and excellent reproducibility.			
Inertsil ODS-SP		The carbon load is controlled to elute highly hydrophobic analytes in shorter time.			
Inertsil ODS-EP		Contains a polar group embedded between an ODS group and silica. Compared with ordinary ODS columns, this column often delivers unique selectivity.			
Inertsil ODS-P		Bonding density of ODS group is very high. As its steric selectivity is quite high, this column is useful for separation of similar compounds.			
Inertsil WP300 C18		It has larger pore on the surface of silica than ordinary columns. High- molecular-weight compounds can be eluted as sharp peaks.			
Column name	Functional group	Features			

Column name	Functional group	Features			
Inertsil C8-4	$\neg \lor \lor \lor$	Inertsil C8-4 is an octyl group (C8) bonded column providing the same separation pattern and extreme inertness to any type of compounds just like Inertsil ODS-4. Recommended as the first choice C8 column.			
Inertsil C8-3	\sim	Bonding density of octyl group is higher than C8-4. Hydrophobic retention is strong among C8 columns.			
Inertsil WP300 C8	\neg	We recommend it as the first choice for high-molecular-weight compounds (MW > 5000). Its pore size is the same as that of			
Inertsil Ph-3		Phenyl group is directly bonded to the silica. Unique selectivity is often delivered, which can be different from that of ODS columns because of the interaction caused by π electron.			
Inertsil Ph-3		delivered, which can be different from that of ODS columns because of			



① Comparison of ODS columns

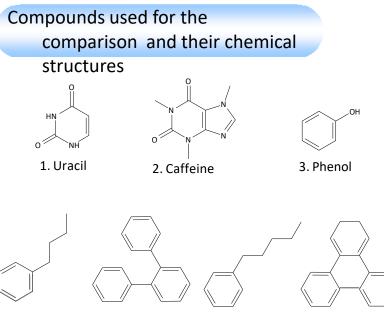
Physical properties of our ODS columns are shown below. The comparison results are in the following pages, where enlarged chromatograms (between 0 min and 4.5 min) are shown on the left side and full scale chromatograms are on the right side.

Column name Stationar	y phase End-capping	Carbon load	Pore size	Surface area	
InertSustain C18	ODS	Yes	14 %	100 Å	350 m²/g
Inertsil ODS-4	ODS	Yes	11 %	100 Å	450 m²/g
Inertsil ODS-3	ODS	Yes	15 %	100 Å	450 m²/g
Inerstil ODS-SP	ODS	Yes	8.5 %	100 Å	450 m²/g
Inertsil ODS-EP	ODS	No	9 %	100 Å	450 m²/g
Inertsil ODS-P	ODS	No	29 %	100 Å	450 m²/g
Inertsil WP300 C18	ODS	Yes	9 %	300 Å	150 m²/g

HPLC condition

Column	: Columns for reversed-phase mode			
	(5 μm, 250 × 4.6 mm I.D.)			
• Eluent	: A) CH ₃ OH B) H ₂ O			
	A/B= 80/20, v/v , 1.0 mL/min			
Temperature	: 40 °C			
Detection	: UV 254 nm			
 Injection Volume 	: 5 μL			

The mobile phase consists of water and methanol. Column dimension and particle size are the same for the all column used in this examination.



4. Butylbenzene 5.



7. Triphenylene

Structures are created using Chemistry 4-D Draw which is provided by ChemInnovation Software, Inc.

The sample solution consists of basic compounds, acidic compounds, alkylbenzens, and polycyclic aromatic hydrocarbons. The retention times and the elution order clarify features of ODS columns.

Sample No.1, Uracil cannot be retained in a reversed phase mode when using the mobile phase condition described on the right. Therefore, Uracil was used to determine the t0 position. t0: Void Volume

Sample No.2, 3, Caffeine and Phenol are used to confirm the amount of residual silanol on the surface of the silica gel. Caffeine elutes later against Phenol when there is a lot of residual

silanol on the surface of the silica gel. *

Sample No.4, *n*-Butylbenzene and Sample No.6, *n*-Amylbenzene were used to determine the hydrophobic property of the column. n-Amylbenzene elu later against *n*-Butylbenzene when the hydrophobicity of the column is high.

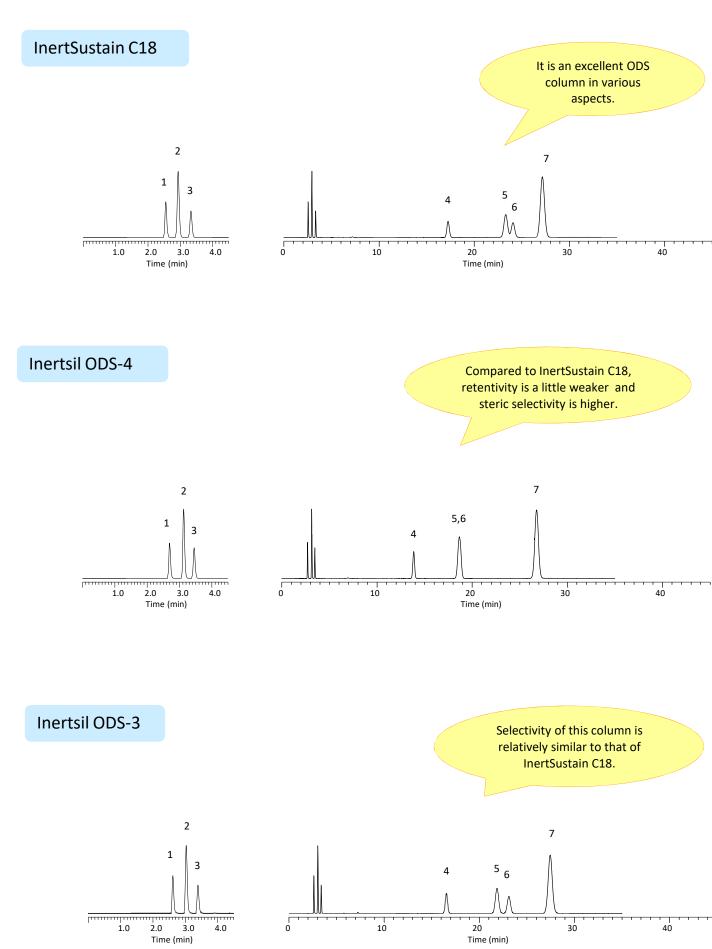
Stereoselectivity is indicated by Sample No.5, *o*-Terphenyl and Sample No.7, Triphenylene. *o*-Terphenyl has a twisted steric structure and

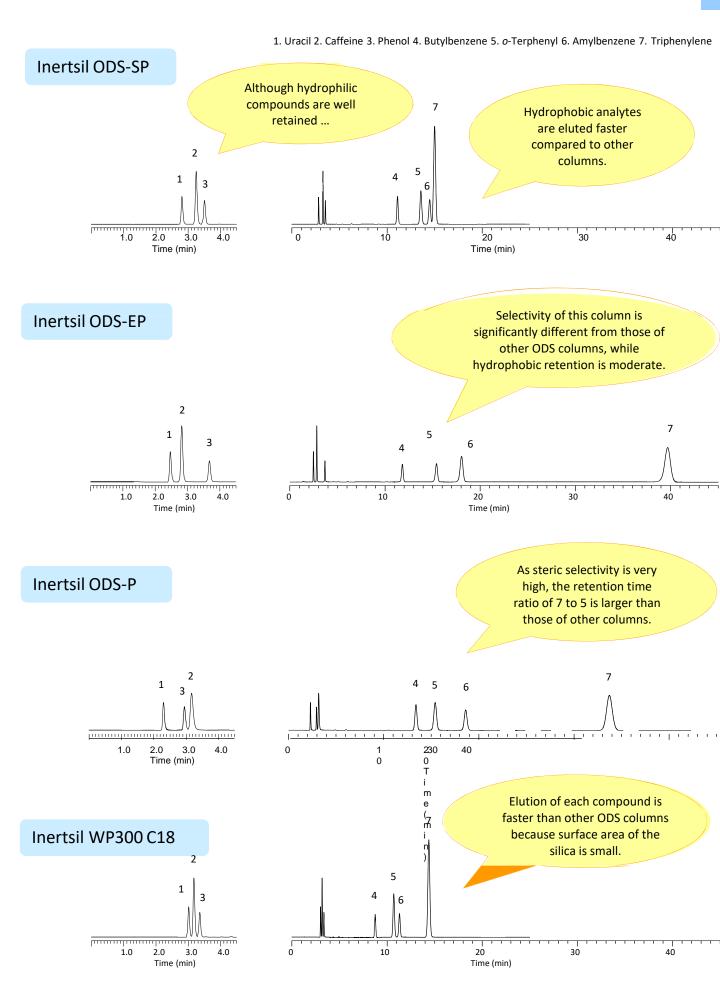
Triphenylene has a planar structure.

Triphenylene elutes later against *o*-Terphenyl when the stereoselectivity of the column is high.

* Reference;
 K. Kimata, K. Iwaguchi, S. Onishi, K. Jinno, R. Eksteen, K. Hosoya,
 M. Araki, N.Tanaka, J. Chromatogr. Sci. 1989, 27, 721-728.

1. Uracil 2. Caffeine 3. Phenol 4. Butylbenzene 5. o-Terphenyl 6. Amylbenzene 7. Triphenylene

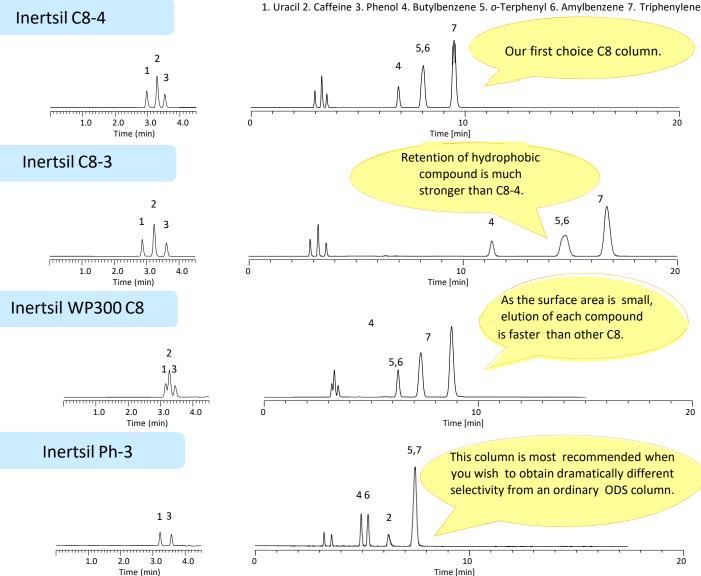




(2) Comparison of C8 and phenyl columns

Physical properties of our C8 and phenyl columns are shown below. The comparison results are also obtained under the same condition as the ODS columns.

Inertsil C8-4 Inertsil	C 8	Y e s	5 %	100 Å	450 m ² /g
C8-3 Inertsil	C 8	Y e s	9 %	100 Å	450 m ² /g
WP300 C8 Inertsil	C8	Y e s	4 %	300 Å	150 m²/g
Ph-3	Ph	No	9.5 %	100 Å	450 m²/g



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